Sandbach School Science Curriculum: Chemistry Combined Impact **Implementation** By the end of Y13, students will know how to: By the end of <u>Y11</u>, students will know: The complex and diverse

Intent

A - Level scientists and have students appreciate that chemistry is fundamental to our world and touches almost every aspect of our existence. We challenge students to think, act and speak like those working in a scientific field would. We do this by using effective questioning techniques in each lesson to push our students to think beyond their first response. Students are expected to carry out practical work in each topic, where it is appropriate, in a responsible manner and record data effectively in order to be able to analyse it and draw conclusions

from it. During practical work, students are expected to select the most appropriate apparatus and justify the choices that they make, thus demonstrating that they are thinking through a problem rather than simply following instructions. Students are expected to consider their own and others safety and independently carry out risk assessments.

Y9-11

To make sure students learn subject content relevant to the qualifications phase exams and community life. To strengthen student confidence in applying their knowledge to exam questions and new situations and being sufficiently adept in transferring of those skills that adequately reflects their understanding of subject and topic content. The qualification phase science curriculum has been structured for the purpose of reinforcing and building upon vocabulary, concepts and visual models studied in the Transition phase Science Curriculum.

Combined 1:15 mins

<u>HT3</u>

C2 Periodic

All 3 Science and HSW

<u>HT4</u>

Biology &

hysics topics

Biology & Physics

<u>HT4</u>

C2 Periodic

table

<u>HT2</u>

C1 Atoms

Review 1: Combined & Triple Nov Y9

Atomic structure only. No extra triple

<u>HT5</u>

7H Atoms, element

& compounds

All 3 Science and HSW

Y7 Review 3: SATs-style past paper questions

<u>HT5</u>

C3

Structure &

bonding

HT1

Bridging

unit 9F

Reactivity

<u>HT6</u>

Biology &

Physics topics

Y7 Review 1 - based on teacher

HT2

7F Acids & Alkalis

assessment from end of module tests

Review 2: Deadline March Y9

Triple: Extra content to be assessed

Combined C2 Periodic table

Y7 Review 2: SATs-style past paper questions

HT6

HSW Topic

Review 3: Deadline June Y9

Y9

HT1

Our School

7E Mixtures &

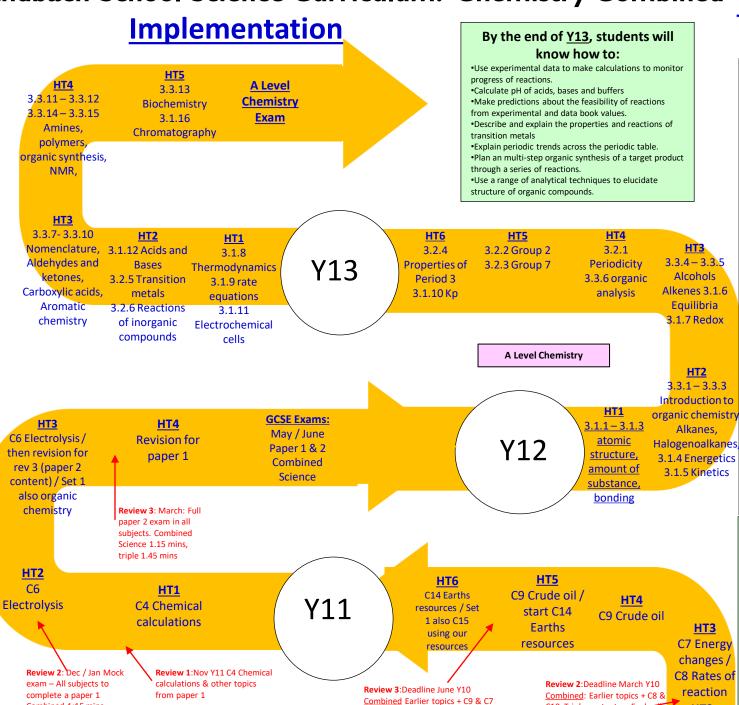
Separation

Combined. P1-2 Content

Triple: P1-3 content

Y7-8 Transition phase

- wonder of science
- recall of scientific knowledge and an ability to apply
- collaborative work
- informed and ethical scientific questions facing society
- responsibility for our planet and the best care for it
- The development of a



Pupils will be taught to: Develop an awe and

- Develop a confident
- scientific concepts Develop the ability to question the credibility
- of reported science Develop confident and independent scientists through individual and
- Allow students to have opinions about the big
- Develop a sense of knowledge to be able to
- curiosity for what else we can learn about the world through science
- Develop transferable and employability skills

at different rates • chemical reactions take place in only three different ways: • proton transfer • electron transfer • electron sharing • energy is conserved in chemical reactions chemical reactions Evaluate risks out the most HT2 atmosphere . independent, Analysis fieldwork and Review 1:Nov C1-3 recap & some safety improvements **HT4** Biology & **Physics topics** results <u>HT3</u> 8F Periodic table data to draw conclusions HT2 8H rocks and hypotheses Evaluate data, showing awareness of potential Y8 Review 1: SATs-style past sources of random and paper questions. All 3 Science and HSW Identify further Key Stage 2 Curriculum their results (International Union of Pure and Applied Chemistry) chemical nomenclature

By the end of transition know how to: Understand that

phenomena of the natural

world can be described in

world can be described in terms of a small number of key ideas in chemistry. These key ideas are of universal application. They underpin many aspects of

the science assessments

Key ideas in chemistry:
• matter is composed of tiny particles called atom and there are about 100 different naturally

occurring types of atoms

elements show periodic

properties

• these periodic properties

can be explained in terms of the atomic structure of

atoms bond by either transferring electrons from one atom to another or by sharing electrons • the shapes of molecules and

the way giant structures are

arranged is of great importance in terms of the way they behave • there are barriers to

reaction so reactions occur at different rates

relationships in their chemical and physical

called elements

the elements

scientific methods and theories develop as earlier explanations are modified to take account of new ideas Make predictions using scientific knowledge and understanding select, plan and carry appropriate types of scientific enquiries to test predictions, including identifying dependent and control Use appropriate techniques, apparatus, and materials during laboratory work, paying attention to health and

Make and record observations and range of methods Evaluate methods and suggest possible Apply mathematical concepts and calculate Present observations

and data using appropriate methods, Interpret observations and data, including identifying patterns and using observations, measurements and Present reasoned explanations, including explaining data in relation to predictions

questions arising from Understand and use SI units and IUPAC

Asking questions Making predictions Setting up tests Observing & Measuring Recording data Interpreting & communicating results

Evaluating

C10 Triple content eg Fuel cett

Y10

HT5

8G metals

Y8 Review 2: SATs-style past

paper questions. All 3

science and HSW

HT6

9E Making materials

Y8 Review 3: SATs-style past

HT1

8E Combustion

paper questions. All 3

Science and HSW

Y8

HT1

C5 Chemical

changes